A-day: Due Fri., May 9 (Assigned Wed, May 7) B-day: Due Mon., May 12 (Assigned Thurs., May 8)

2008 Magnetism 2

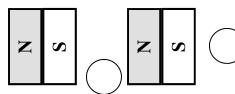
4.

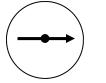
9. A. Label N and S on

each of the magnets.

B. Fill in the compasses.

- 1. What variable do we use for magnetic field?
- 2. A. Draw the magnetic field between the magnets.B. Fill in the compasses.





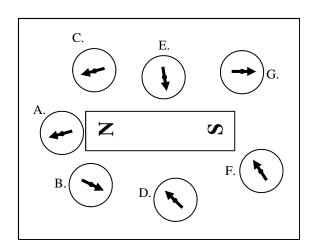
A. Label N and S on the compass.
 B. Label red and black on the compass.

Label N and S on the

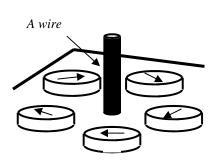
N S

horseshoe magnet.

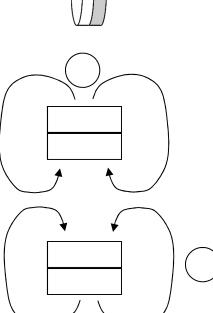
- 5. What symbol do we use for into the page?
- 6. What symbol do we use for out of the page?
- 7. A. Draw B between the donut magnets.B. What will happen to the small magnet between the magnets?
- 8. Which of the compasses below are correct?

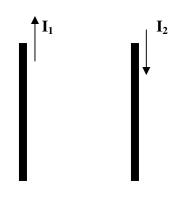


10. A. Is the current in the wire going up or down?B. On the left side of the wire, is B into or out of the page?



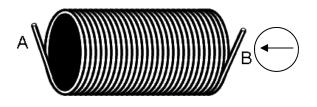
- 11. A. Draw B on both sides of each wire at the right (*draw it close to each wire*).
 - B. In between the two wires are the two magnetic fields go ing the same direction or opposite directions?
 - C. Will the wires be attracted or repelled by each other?





2008 Magnetism 2

- 12. Fill in the compass.
 13. A. If + current goes in side 1, which side of the solenoid is N?
 B. Fill in the compasses.
 C. Which side is hooked up to the negative side of the battery?
- 14. A. Label N and S on the solenoid.
 - B. Which side did positive current go in?





15. If the small circle at the left is a wire, and the arrow shows the direction of the magnetic field, is the current coming out of or into the page?

- 16. The X shows the direction of current in a wire.A. Fill in the three compasses.
 - B. Draw N and S on the magnet.

Review for final.

- 17. At what temperature does water turn to ice?
- 18. At what temperature does water turn to steam?
- 19. 5 kg of water is originally at -5° C and is heated up to 60° C.
 - A. What state of matter was the water at originally: solid, liquid, or gas?
 - B. What state of matter did water end up as?
 - C. How many degrees did the water increase in its first state?
 - D. How many degrees did the water increase in its second state?
 - E. Did the water under go fusion or vaporization?
 - F. Using your notes on "Heat" and "Latent Heat", calculate the total heat necessary to raise the water's temperature.