

**A-Day: Due Mon., Jan 8 (Assigned: 1/4)**  
**B-Day: Due Tues., Jan 9 (Assigned: 12/6)**

## Heat 1

- 1) Conduction (1), Convection (2), or Radiation (3)?
  - A. \_\_\_ Your hand gets warm while underneath (but not touching) a hot pot of water.
  - B. \_\_\_ Your hand cools down when pushed against the metal on your desk.
  - C. \_\_\_ Why smoke rises above a campfire.
  - D. \_\_\_ Molecules bumping against each other.
- 2) Use this picture to answer the following:

30°C	60°C
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  - A. Use an arrow to show which way heat will move.
  - B. Is the object at thermal equilibrium?
  - C. Which object will lose heat?
  - D. Which object will gain heat?
  - E. Which object has a negative  $\Delta Q$ ?
  - F. Which object has a positive  $\Delta Q$ ?
  - G. How does the amount of  $Q$  gained or lost compare?
- 3) From the lab:
  - A. Which dropped temperature faster when put into the room temperature water: the tin can or the glass beaker?
  - B. Why?
  
  - C. Which gained temperature faster in the heat lamp: the black can or the white can?
  - D. Why?
- 4) You put ice into a cup of hot chocolate. The ice gives its cold to the liquid. Yes or no and why?
- 5) An object is at 35°C. Convert this temperature to the Kelvin scale.
- 6) Convert 80°F to Celsius.
- 7) How much heat is necessary to raise 15 kg of water from 20°C to 40°C? ( $C_p$  of water = 4186)
  
- 8) Using the same mass and temperature change as before for copper ( $C_p$  of copper = 387).
  
- 9) How much heat is necessary to boil 15 kg of water? ( $L_v$  of water =  $2.26 \times 10^6$ )
  
- 10) Use Q7-9 to answer the following:
  - A. Does it take more heat to raise the temperature of copper or water?
  - B. Did it take more heat to raise the temperature of water or to change it to a gas?
  - C. So, if the  $c_p$  of iron = 448 and  $c_p$  of aluminum = 899, which one will require the most  $Q$  to change its temperature?
- 11) Which part of the desk feels colder: the metal or the wood?
- 12) Which one is actually colder: the metal or the wood?
- 13) Why do they feel different?
  
- 14) Does  $Q$  have to go into or out of water as it changes from liquid to gas?
- 15) When your hand is wet and you blow on it, does your hand feel hot or cold?
- 16) So, when water freezes, does it give off heat or absorb heat?
- 17) Why, then, do citrus growers in Florida spray water on their fruit to protect it from freezing weather?

*More on back*

Fill in the blanks to help you better understand states of matter vocabulary. Remember that there may not be the correct number of blanks for each vocabulary word.

