## 2009 PreAP Thermo 2

- 1) Isothermal (T); Isovolumetric (V); Adiabatic (A)?
  - A. \_\_\_\_\_A tire being rapidly inflated.
  - B. \_\_\_\_\_A tire expanding gradually as it is heated.
  - C. \_\_\_\_\_A tire being heated with it is in a rigid metal container.
  - D. \_\_\_\_ In a refrigerator when the compressor compresses the refrigerant quickly.
  - E. \_\_\_\_ In a refrigerator when the refrigerant (which is in a metal tube) absorbs heat from the inside of the refrigerator.
  - F. \_\_\_\_ In a refrigerator when the refrigerant expands quickly.
  - G. \_\_\_\_Q =W.
  - H.  $\_\_\Delta U = 0.$

I. \_\_\_\_\_ $\Delta U = Q$ .

- 2) Positive, Negative, or Zero?
  - A.  $\_\_\Delta U$  during an isovolumetric process if heat is removed.
  - B. \_\_\_\_Q in an isovolumetric process if  $\Delta U$  is negative.
  - C.  $\_\_\Delta U$  during an isothermal process.
  - D. \_\_\_\_Q in an adiabatic process if the gas expands.
  - E.\_\_\_\_  $\Delta U$  if Q =W.
  - F.\_\_\_\_  $\Delta U$  when positive work is done on the gas (Q = 0).
  - G. \_\_\_\_\_  $\Delta U$  when negative work is done by the gas (Q = 0).
  - H. \_\_\_\_  $\Delta U$  during an adiabatic expansion.
  - I. \_\_\_\_\_ Work done by the gas when Q = 0, but temperature decreases.
  - J. \_\_\_\_\_ Work done by the gas when the volume of the gas increases.
  - K. \_\_\_\_Q in an isothermal process if the gas compresses.

  - L.  $\Delta U$  if Q = 0 and the gas is compressed. M. Work done by the gas during an isovolumetric process.
- 3) An engine gains  $2.56 \times 10^7$  J of energy from combustion. If the engine expels  $1.15 \times 10^7$  J, how efficient is the engine?
- 4) If a refrigerator is left open in the middle of a room, does the room's overall temperature increase or decrease over time? (Support your answer, of course.)
- 5) Book Ch. 11 p. 431: Q 8, 9, 10, 11, 13, 14, 20, 28.