Name: Period:	HW—25:ATL — After Test Mr. Murray, IPC www.aisd.net/smurray	Assigned: Mon., 12/15/03 Due: Wed., 12/17/03
1. Make balanced ionic compounds from:	3. An element has 20 protons and a mass number of 46.	4. Find the molecular mass of CO ₃
Li and F $\label{eq:mg} \mbox{Mg and Cl}$ $\mbox{Na and CrO}_4^{\ 2\mbox{-}}$	Mass #: 46	5. A 30 gram object is dropped into a graduated cylinder. The water in the cylinder rises from 10 mL to 15 mL. Find the object's density.
	# of protons:	
2. Name these: Li_2O ; $Mg(CO_3)_2$; S_4Cl_2	# of neutrons: # of valence electrons:	6. On the back draw a density column of: liquid A (1.0 g/mL); liquid B (0.75 g/mL); liquid C (2.4 g/mL); rock (3.5 g/mL) And label what you know.
	# of full electrons shells:	
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