Name:

Period:

1. An element has 15 protons and a mass number of 32.

Mass #: 32

of protons:

of neutrons:

Name:

of valence electrons:

of full electrons shells:

HW—25:AT — Review for Test Mr. Murray, IPC www.aisd.net/smurray

2. Make balanced ionic compounds from:

3. Name these: MgO; C_2F_4 Mg(NO₃)₂;

HW—25:AT — Review for Test

Mr. Murray, IPC

Ca and F

Mg and S

Al and NO₃¹⁻

Assigned: Mon., 12/15/03 Due: Wed., 12/17/03

4. A 20 gram object is dropped into a graduated cylinder. The water in the cylinder rises from 20 mL to 30 mL. Find the object's density.

5. Find the molecular mass of CO_2

6. On the back draw a density column of: liquid A (2.3 g/mL); liquid B (1.0 g/mL); liquid C (0.87 g/mL); object 1 (1.5 g/mL)

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