Name:	HW—20:1	Assigned: Thurs., 11/6/03
Period:	Physical vs. Chemical Changes Mr. Murray, IPC	Due: Mon., 11/10/03
Physical or Chemical Change?	Why? (Your evidence)	5. $2KClO_3 \rightarrow 2KCl + 3O_2$
1. Cutting Paper		Circle the products. Underline the reactants.
2. Mixing two liquids makes bubbles		
3. Mixing Coolaid		How many oxygens are on the reactant side?
		How many oxygens are on the product side?
4. Wood Rotting		- 7. How many total atoms does 3(NO <sub>3</sub> ) have?
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Period:	Mr. Murray, IPC	<b>Duc.</b> 11011., 11/10/00
Physical or Chemical Change?	Why? (Your evidence)	5. $2KClO_3 \rightarrow 2KCl + 3O_2$
1. Cutting Paper		Circle the products. Underline the reactants.
2. Mixing two liquids makes bubbles		6. $Al_2(CrO_4)_3 + 6NaCl \rightarrow 3Na_2(CrO_4) + 2AlCl_3$
3. Mixing Coolaid		How many oxygens are on the reactant side?
		How many oxygens are on the product side?

7. How many total atoms does 3(NO<sub>3</sub>) have?

4. Wood Rotting\_\_\_\_\_