Name: _____

Period: _____

HW—20:1
Physical vs. Chemical Changes
Mr. Murray, IPC

Assigned: Thurs., 11/6/03
Due: Mon., 11/10/03

Physical or Chemical Change?	Why? (Your evidence)
1. Cutting Paper _____	_____
2. Mixing two liquids makes bubbles _____	_____
3. Mixing Coolaid _____	_____
4. Wood Rotting_____	_____

5. $2KClO_3 \rightarrow 2KCl + 3O_2$

Circle the products.
Underline the reactants.

6. $Al_2(CrO_4)_3 + 6NaCl \rightarrow 3Na_2(CrO_4) + 2AlCl_3$

How many oxygens are on the reactant side?

How many oxygens are on the product side?

7. How many total atoms does $3(NO_3)$ have?

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