Name:	HW—20:1L	Assigned: Mon., 11/10/03
Period:	Physical vs. Chemical Changes Mr. Murray, IPC	Due: Wed., 11/12/03
Physical or Chemical Change?	Why? (Your evidence)	5. $K(NO_3) \rightarrow K + NO_3$
1. Burning Paper		Circle the products. Underline the reactants.
2. Mixing two liquids and they change color		
3. Boiling water		— How many oxygens are on the reactant side?
4.0		How many magnesiums are on the product side?
4. Sawing wood		7. How many total atoms does 4(NO ₃) have?
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1. Burning Paper		Circle the products. Underline the reactants.
2. Mixing two liquids and they change color		- 6. Mg ₃ (PO ₄) ₂ +6NaCl → 2Na ₃ (PO ₄)+3MgCl ₂
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