

Name: _____

Period: _____

HW—19:6L Naming and Covalent Compounds

Mr. Murray, IPC
www.aisd.net/smurray

Assigned: Wedn., 10/29/03

Due: Fri., 10/31/03

Ionic, Covalent, or
Polyatomic (I, P, or C)

Name

5. Give the ionic notation for
an element with 3 protons and
2 electrons. Cation or anion?

7. Draw the covalent mole-
cule of Cl₂.

1. BeO _____

2. C₂O₄ _____

3. OCl₃ _____

4. Na₂SO₄ _____

6. Make a compound from
Mg and N.

8. Using dot diagrams draw
Oxygen Dichloride. (Use the
back if you need more space)

Name: _____

Period: _____

HW—19:6L Naming and Covalent Compounds

Mr. Murray, IPC
www.aisd.net/smurray

Assigned: Wedn., 10/29/03

Due: Fri., 10/31/03

Ionic, Covalent, or
Polyatomic (I, P, or C)

Name

5. Give the ionic notation for
an element with 3 protons and
2 electrons. Cation or anion?

7. Draw the covalent mole-
cule of Cl₂.

1. BeO _____

2. C₂O₄ _____

3. OCl₃ _____

4. Na₂SO₄ _____

6. Make a compound from
Mg and N.

8. Using dot diagrams draw
Oxygen Dichloride. (Use the
back if you need more space)

Name: _____

Period: _____

HW—19:6L Naming and Covalent Compounds

Mr. Murray, IPC
www.aisd.net/smurray

Assigned: Wedn., 10/29/03

Due: Fri., 10/31/03

Ionic, Covalent, or
Polyatomic (I, P, or C)

Name

5. Give the ionic notation for
an element with 3 protons and
2 electrons. Cation or anion?

7. Draw the covalent mole-
cule of Cl₂.

1. BeO _____

2. C₂O₄ _____

3. OCl₃ _____

4. Na₂SO₄ _____

6. Make a compound from
Mg and N.

8. Using dot diagrams draw
Oxygen Dichloride. (Use the
back if you need more space)