Name: Period:	Ovidation Numbers		Assigned: Fri., 10/17/03 Due: Tues., 10/21/03
1. Metal or non-metal?	3. Give the ion notation for Fluorine that gains 1 electron.	5. Will these form ionic compounds?:	7. Covalent or Ionic? Li ₂ O
K		Ca $^{2+}$ and O $^{2-}$	
P			
2. Show the dot diagram for:	4. Give the ion notation for an atom with 9 protons and 10 electrons.	6. Find oxidation numbers for:	Make a balance compound of Ca and F:
Sulfur		Cl:	
		Be: Al:	
Name:	HW—19:2L Oxidation Numbers		Assigned: Fri., 10/17/03
Period:	- Mr. Muri		Due: Tues., 10/21/03
1. Metal or non-metal?	3. Give the ion notation for Fluorine that gains 1 electron.	5. Will these form ionic compounds?:	7. Covalent or Ionic? Li ₂ O
K		Ca $^{2+}$ and O $^{2-}$	
P			
	4. Give the ion notation for an atom with 9 protons and 10 electrons.	6. Find oxidation numbers for:	Make a balance compound of Ca and F:
2. Show the dot diagram for: Sulfur		Cl:	
		Be:	
		Al:	
Name:	HW—J	19:2L	Assigned: Fri., 10/17/03
Period:	Ovidation Numbers		Due: Tues., 10/21/03
1. Metal or non-metal?	3. Give the ion notation for Fluorine that gains 1 electron.	5. Will these form ionic compounds?:	7. Covalent or Ionic? Li ₂ O
K		Ca $^{2+}$ and O $^{2-}$	
P			
2. Show the dot diagram for: Sulfur	4. Give the ion notation for an atom with 9 protons and 10 electrons.	6. Find oxidation numbers for:	
		Cl:	Make a balance compound of Ca and F:
		Be:	
		Al:	

vame:	
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